

Publication

A counterion study of a series of $[\text{Cu}(\text{P}^*\text{P})(\text{N}^*\text{N})][\text{A}]$ compounds with bis(phosphane) and 6-methyl and 6,6'-dimethyl-substituted 2,2'-bipyridine ligands for light-emitting electrochemical cells

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The syntheses and characterisations of a series of heteroleptic copper(i) compounds $[\text{Cu}(\text{POP})(\text{Mebpy})][\text{A}]$, $[\text{Cu}(\text{POP})(\text{Me}(2)\text{bpy})][\text{A}]$, $[\text{Cu}(\text{xantphos})(\text{Mebpy})][\text{A}]$ and $[\text{Cu}(\text{xantphos})(\text{Me}(2)\text{bpy})][\text{A}]$ in which $[\text{A}](-)$ is $[\text{BF}_4](-)$, $[\text{PF}_6](-)$, $[\text{BPh}_4](-)$ and $[\text{BAr}_4\text{F}](-)$ (Mebpy = 6-methyl-2,2'-bipyridine, $\text{Me}(2)\text{bpy}$ = 6,6'-dimethyl-2,2'-bipyridine, POP = oxydi(2,1-phenylene)bis(diphenylphosphane), xantphos = (9,9-dimethyl-9H-xanthene-4,5-diyl)bis(diphenylphosphane), $[\text{BAr}_4\text{F}](-)$ = tetrakis(3,5-bis(trifluoromethyl)phenyl)borate) are reported. Nine of the compounds have been characterised by single crystal X-ray crystallography, and the consequences of the different anions on the packing interactions in the solid state are discussed. The effects of the counterion on the photophysical properties of $[\text{Cu}(\text{POP})(\text{N}^<\text{N})][\text{A}]$ and $[\text{Cu}(\text{xantphos})(\text{N}^<\text{N})][\text{A}]$ ($\text{N}^<\text{N}$ = Mebpy and $\text{Me}(2)\text{bpy}$) have been investigated. In the solid-state emission spectra, the highest energy emission maxima are for $[\text{Cu}(\text{xantphos})(\text{Mebpy})][\text{BPh}_4]$ and $[\text{Cu}(\text{xantphos})(\text{Me}(2)\text{bpy})][\text{BPh}_4]$ (λ_{emmax} = 520 nm) whereas the lowest energy λ_{emmax} values occur for $[\text{Cu}(\text{POP})(\text{Mebpy})][\text{PF}_6]$ and $[\text{Cu}(\text{POP})(\text{Mebpy})][\text{BPh}_4]$ (565 nm and 563 nm, respectively). Photoluminescence quantum yields (PLQYs) are noticeably affected by the counterion; in the $[\text{Cu}(\text{xantphos})(\text{Me}(2)\text{bpy})][\text{A}]$ series, solid-state PLQY values decrease from 62% for $[\text{PF}_6](-)$, to 44%, 35% and 27% for $[\text{BF}_4](-)$, $[\text{BPh}_4](-)$ and $[\text{BAr}_4\text{F}](-)$, respectively. This latter series of compounds was used as active electroluminescent materials on light-emitting electrochemical cells (LECs). The luminophores were mixed with ionic liquids (ILs) $[\text{EMIM}][\text{A}]$ ($[\text{EMIM}](+) = [1\text{-ethyl-3-methylimidazolium}](+)$) containing the same or different counterions than the copper(i) complex. LECs containing $[\text{Cu}(\text{xantphos})(\text{Me}(2)\text{bpy})][\text{BPh}_4]$ and $[\text{Cu}(\text{xantphos})(\text{Me}(2)\text{bpy})][\text{BAr}_4\text{F}]$ failed to turn on under the LEC operating conditions, whereas those with the smaller $[\text{PF}_6](-)$ or $[\text{BF}_4](-)$ counterions had rapid turn-on times and exhibited maximum luminances of 173 and 137 cd m^{-2} and current efficiencies of 3.5 and 2.6 cd A^{-1} , respectively, when the IL contained the same counterion as the luminophore. Mixing the counterions ($[\text{PF}_6](-)$ and $[\text{BF}_4](-)$) of the active complex and the IL led to a reduction in all the figures of merit of the LECs.

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