

Publication

2H-fractionations during the biosynthesis of carbohydrates and lipids imprint a metabolic signal on the $\delta^2\text{H}$ values of plant organic compounds.

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Hydrogen (H) isotope ratio ($\delta^2\text{H}$) analyses of plant organic compounds have been applied to assess ecohydrological processes in the environment despite a large part of the $\delta^2\text{H}$ variability observed in plant compounds not being fully elucidated. We present a conceptual biochemical model based on empirical H isotope data that we generated in two complementary experiments that clarifies a large part of the unexplained variability in the $\delta^2\text{H}$ values of plant organic compounds. The experiments demonstrate that information recorded in the $\delta^2\text{H}$ values of plant organic compounds goes beyond hydrological signals and can also contain important information on the carbon and energy metabolism of plants. Our model explains where $\delta^2\text{H}$ -fractionations occur in the biosynthesis of plant organic compounds and how these $\delta^2\text{H}$ -fractionations are tightly coupled to a plant's carbon and energy metabolism. Our model also provides a mechanistic basis to introduce H isotopes in plant organic compounds as a new metabolic proxy for the carbon and energy metabolism of plants and ecosystems. Such a new metabolic proxy has the potential to be applied in a broad range of disciplines, including plant and ecosystem physiology, biogeochemistry and palaeoecology.

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